

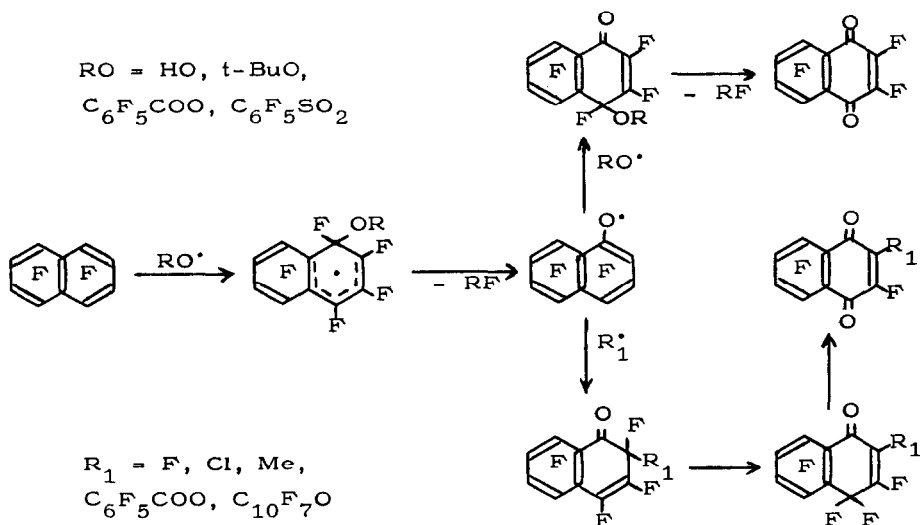
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THE INTERACTION OF POLYFLUOROAROMATIC COMPOUNDS WITH OXYGEN CONTAINING RADICALS

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Addition of oxygen containing radicals (HO^\bullet , $t\text{-BuO}^\bullet$, $\text{C}_6\text{F}_5\text{COO}^\bullet$, $\text{C}_6\text{F}_5\text{SO}_2^\bullet$) to octafluoronaphthalene occurs mainly in the 1-position. The radical intermediates are stabilised by formation of heptafluoronaphthoxyl radical, further transformations of which lead mainly to polyfluorinated derivatives of 1-oxodihydronaphthalenes and 1,4-naphthoquinone. Reactions with HO^\bullet may involve aromatic ring cleavage both in octafluoronaphthalene and heptafluoronaphthols.



Hexafluorobenzene reacts with hydroxyl radical to give pentafluorophenol. In pentafluorodiphenyl, hydroxylation of the non-fluorinated nucleus takes place. Interaction of hydrogen peroxide with pentafluorophenol leads, depending on the reaction conditions, to the formation of either tetrakis-(pentafluorophenoxy)-1,4-benzoquinone or difluoromaleic acid.